

- \*P19.1 (a) The mass of an average neutral hydrogen atom is 1.007 9u. Losing one electron reduces its mass by a negligible amount, to

$$1.007\ 9(1.660 \times 10^{-27}\ \text{kg}) - 9.11 \times 10^{-31}\ \text{kg} = \boxed{1.67 \times 10^{-27}\ \text{kg}}.$$

Its charge, due to loss of one electron, is

$$0 - 1(-1.60 \times 10^{-19}\ \text{C}) = \boxed{+1.60 \times 10^{-19}\ \text{C}}.$$

- (b) By similar logic, charge =  $\boxed{+1.60 \times 10^{-19}\ \text{C}}$

$$\text{mass} = 22.99(1.66 \times 10^{-27}\ \text{kg}) - 9.11 \times 10^{-31}\ \text{kg} = \boxed{3.82 \times 10^{-26}\ \text{kg}}$$

- (c) charge of  $\text{Cl}^- = \boxed{-1.60 \times 10^{-19}\ \text{C}}$

$$\text{mass} = 35.453(1.66 \times 10^{-27}\ \text{kg}) + 9.11 \times 10^{-31}\ \text{kg} = \boxed{5.89 \times 10^{-26}\ \text{kg}}$$

- (d) charge of  $\text{Ca}^{++} = -2(-1.60 \times 10^{-19}\ \text{C}) = \boxed{+3.20 \times 10^{-19}\ \text{C}}$

$$\text{mass} = 40.078(1.66 \times 10^{-27}\ \text{kg}) - 2(9.11 \times 10^{-31}\ \text{kg}) = \boxed{6.65 \times 10^{-26}\ \text{kg}}$$

- (e) charge of  $\text{N}^{3-} = 3(-1.60 \times 10^{-19}\ \text{C}) = \boxed{-4.80 \times 10^{-19}\ \text{C}}$

$$\text{mass} = 14.007(1.66 \times 10^{-27}\ \text{kg}) + 3(9.11 \times 10^{-31}\ \text{kg}) = \boxed{2.33 \times 10^{-26}\ \text{kg}}$$

- (f) charge of  $\text{N}^{4+} = 4(1.60 \times 10^{-19}\ \text{C}) = \boxed{+6.40 \times 10^{-19}\ \text{C}}$

$$\text{mass} = 14.007(1.66 \times 10^{-27}\ \text{kg}) - 4(9.11 \times 10^{-31}\ \text{kg}) = \boxed{2.32 \times 10^{-26}\ \text{kg}}$$

- (g) We think of a nitrogen nucleus as a seven-times ionized nitrogen atom.

$$\text{charge} = 7(1.60 \times 10^{-19}\ \text{C}) = \boxed{1.12 \times 10^{-18}\ \text{C}}$$

$$\text{mass} = 14.007(1.66 \times 10^{-27}\ \text{kg}) - 7(9.11 \times 10^{-31}\ \text{kg}) = \boxed{2.32 \times 10^{-26}\ \text{kg}}$$

- (h) charge =  $\boxed{-1.60 \times 10^{-19}\ \text{C}}$

$$\text{mass} = [2(1.007\ 9) + 15.999]1.66 \times 10^{-27}\ \text{kg} + 9.11 \times 10^{-31}\ \text{kg} = \boxed{2.99 \times 10^{-26}\ \text{kg}}$$

P19.2 (a) 
$$N = \left( \frac{10.0\ \text{grams}}{107.87\ \text{grams/mol}} \right) \left( 6.02 \times 10^{23}\ \frac{\text{atoms}}{\text{mol}} \right) \left( 47\ \frac{\text{electrons}}{\text{atom}} \right) = \boxed{2.62 \times 10^{24}}$$

(b) # electrons added =  $\frac{Q}{e} = \frac{1.00 \times 10^{-3}\ \text{C}}{1.60 \times 10^{-19}\ \text{C/electron}} = 6.25 \times 10^{15}$

or  $\boxed{2.38\ \text{electrons for every } 10^9\ \text{already present}}.$

P19.3 If each person has a mass of  $\approx 70$  kg and is (almost) composed of water, then each person contains

$$N = \left( \frac{70\,000 \text{ grams}}{18 \text{ grams/mol}} \right) \left( 6.02 \times 10^{23} \frac{\text{molecules}}{\text{mol}} \right) \left( 10 \frac{\text{protons}}{\text{molecule}} \right) = 2.3 \times 10^{28} \text{ protons.}$$

With an excess of 1% electrons over protons, each person has a charge

$$q = 0.01(1.6 \times 10^{-19} \text{ C})(2.3 \times 10^{28}) = 3.7 \times 10^7 \text{ C.}$$

$$\text{So } F = k_e \frac{q_1 q_2}{r^2} = (9 \times 10^9) \frac{(3.7 \times 10^7)^2}{0.6^2} \text{ N} = 4 \times 10^{25} \text{ N} \left[ \sim 10^{26} \text{ N} \right].$$

This force is almost enough to lift a weight equal to that of the Earth:

$$Mg = 6 \times 10^{24} \text{ kg}(9.8 \text{ m/s}^2) = 6 \times 10^{25} \text{ N} \sim 10^{26} \text{ N.}$$

P19.4 The force on one proton is  $\vec{F} = \frac{k_e q_1 q_2}{r^2}$  away from the other proton. Its magnitude is

$$(8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2) \left( \frac{1.6 \times 10^{-19} \text{ C}}{2 \times 10^{-15} \text{ m}} \right)^2 = \boxed{57.5 \text{ N}}.$$

$$\text{P19.5 } F_1 = k_e \frac{q_1 q_2}{r^2} = \frac{(8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2)(7.00 \times 10^{-6} \text{ C})(2.00 \times 10^{-6} \text{ C})}{(0.500 \text{ m})^2} = 0.503 \text{ N}$$

$$F_2 = k_e \frac{q_1 q_2}{r^2} = \frac{(8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2)(7.00 \times 10^{-6} \text{ C})(4.00 \times 10^{-6} \text{ C})}{(0.500 \text{ m})^2} = 1.01 \text{ N}$$

$$F_x = 0.503 \cos 60.0^\circ + 1.01 \cos 60.0^\circ = 0.755 \text{ N}$$

$$F_y = 0.503 \sin 60.0^\circ - 1.01 \sin 60.0^\circ = -0.436 \text{ N}$$

$$\vec{F} = (0.755 \text{ N})\hat{i} - (0.436 \text{ N})\hat{j} = \boxed{0.872 \text{ N at an angle of } 330^\circ}$$

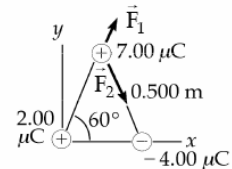


FIG. P19.5

P19.7 (a) The force is one of attraction. The distance  $r$  in Coulomb's law is the distance between centers. The magnitude of the force is

$$F = \frac{k_e q_1 q_2}{r^2} = (8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2) \frac{(12.0 \times 10^{-9} \text{ C})(18.0 \times 10^{-9} \text{ C})}{(0.300 \text{ m})^2} = \boxed{2.16 \times 10^{-5} \text{ N}}.$$

(b) The net charge of  $-6.00 \times 10^{-9} \text{ C}$  will be equally split between the two spheres, or  $-3.00 \times 10^{-9} \text{ C}$  on each. The force is one of repulsion, and its magnitude is

$$F = \frac{k_e q_1 q_2}{r^2} = (8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2) \frac{(3.00 \times 10^{-9} \text{ C})(3.00 \times 10^{-9} \text{ C})}{(0.300 \text{ m})^2} = \boxed{8.99 \times 10^{-7} \text{ N}}.$$

\*P19.8 Let the third bead have charge  $Q$  and be located distance  $x$  from the left end of the rod. This bead will experience a net force given by

$$\vec{F} = \frac{k_e(3q)Q}{x^2} \hat{i} + \frac{k_e(q)Q}{(d-x)^2} (-\hat{i}).$$

The net force will be zero if  $\frac{3}{x^2} = \frac{1}{(d-x)^2}$ , or  $d-x = \frac{x}{\sqrt{3}}$ .

This gives an equilibrium position of the third bead of  $x = \boxed{0.634d}$ .

The equilibrium is .

P19.10 For equilibrium,  $\vec{F}_e = -\vec{F}_g$

or  $q\vec{E} = -mg(-\hat{j}).$

Thus,  $\vec{E} = \frac{mg}{q} \hat{j}.$

(a)  $\vec{E} = \frac{mg}{q} \hat{j} = \frac{(9.11 \times 10^{-31} \text{ kg})(9.80 \text{ m/s}^2)}{(-1.60 \times 10^{-19} \text{ C})} \hat{j} = \boxed{-(5.58 \times 10^{-11} \text{ N/C}) \hat{j}}$

(b)  $\vec{E} = \frac{mg}{q} \hat{j} = \frac{(1.67 \times 10^{-27} \text{ kg})(9.80 \text{ m/s}^2)}{(1.60 \times 10^{-19} \text{ C})} \hat{j} = \boxed{(1.02 \times 10^{-7} \text{ N/C}) \hat{j}}$

- P19.11** The point is designated in the sketch. The magnitudes of the electric fields,  $E_1$ , (due to the  $-2.50 \times 10^{-6}$  C charge) and  $E_2$  (due to the  $6.00 \times 10^{-6}$  C charge) are

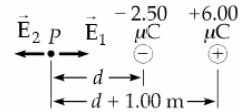


FIG. P19.11

$$E_1 = \frac{k_e q}{r^2} = \frac{(8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2)(2.50 \times 10^{-6} \text{ C})}{d^2} \quad (1)$$

$$E_2 = \frac{k_e q}{r^2} = \frac{(8.99 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2)(6.00 \times 10^{-6} \text{ C})}{(d + 1.00 \text{ m})^2} \quad (2)$$

Equate the right sides of (1) and (2)

$$\text{to get} \quad (d + 1.00 \text{ m})^2 = 2.40d^2$$

$$\text{or} \quad d + 1.00 \text{ m} = \pm 1.55d$$

$$\text{which yields} \quad d = 1.82 \text{ m}$$

$$\text{or} \quad d = -0.392 \text{ m}.$$

The negative value for  $d$  is unsatisfactory because that locates a point between the charges where both fields are in the same direction.

Thus,  $d = \boxed{1.82 \text{ m to the left of the } -2.50 \mu\text{C charge}}.$

**P19.13** (a) 
$$\vec{E}_1 = \frac{k_e |q_1|}{r_1^2} (-\hat{j}) = \frac{(8.99 \times 10^9)(3.00 \times 10^{-9})}{(0.100)^2} (-\hat{j}) = -(2.70 \times 10^3 \text{ N/C})\hat{j}$$

$$\vec{E}_2 = \frac{k_e |q_2|}{r_2^2} (-\hat{i}) = \frac{(8.99 \times 10^9)(6.00 \times 10^{-9})}{(0.300)^2} (-\hat{i}) = -(5.99 \times 10^2 \text{ N/C})\hat{i}$$

$$\vec{E} = \vec{E}_2 + \vec{E}_1 = \boxed{-(5.99 \times 10^2 \text{ N/C})\hat{i} - (2.70 \times 10^3 \text{ N/C})\hat{j}}$$

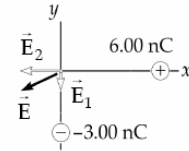


FIG. P19.13

(b) 
$$\vec{F} = q\vec{E} = (5.00 \times 10^{-9} \text{ C})(-599\hat{i} - 2700\hat{j}) \text{ N/C}$$

$$\vec{F} = (-3.00 \times 10^{-6} \hat{i} - 13.5 \times 10^{-6} \hat{j}) \text{ N} = \boxed{(-3.00\hat{i} - 13.5\hat{j}) \mu\text{N}}$$