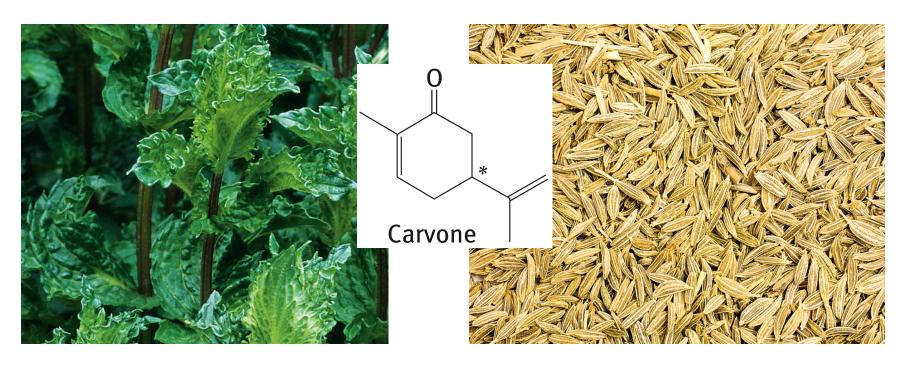
Chapter 5: Stereoisomerism

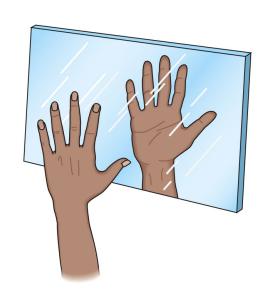
Stereoisomers are compounds that have the same structural formula in terms of order of attachment, but differ in arrangements of the atoms in space.



The difference in odor between caraway seeds and mint leaves arises from two stereoisomers of carvone due to different arrangement of atoms at the carbon (*)

Mirror-image relationship of chiral and achiral objects

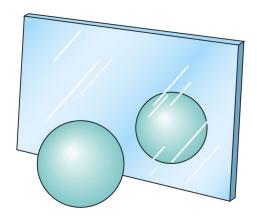
The mirror image of a left hand is not a left hand, but a right hand.





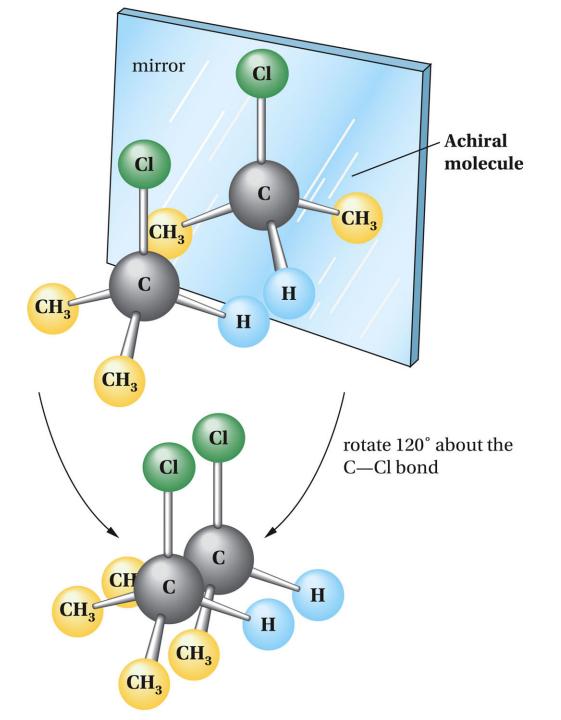
Chiral object

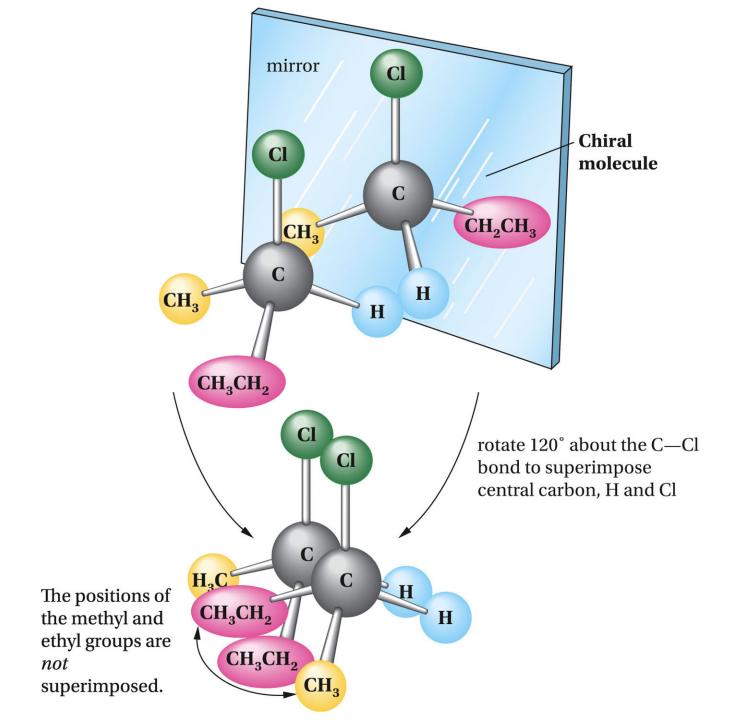
The mirror image of a ball is identical with the object itself.





Achiral object

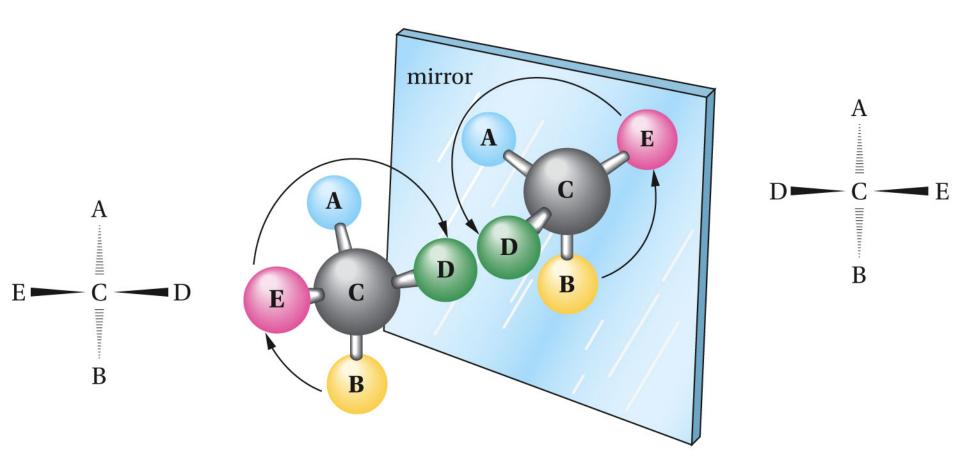


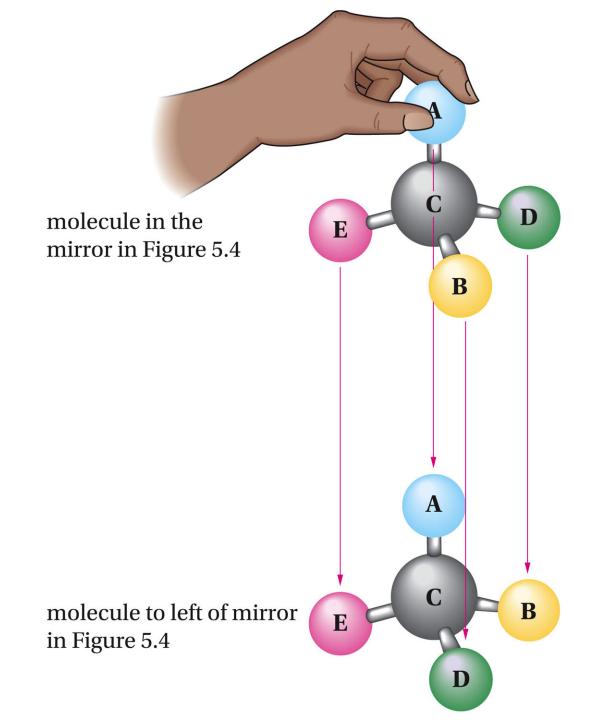


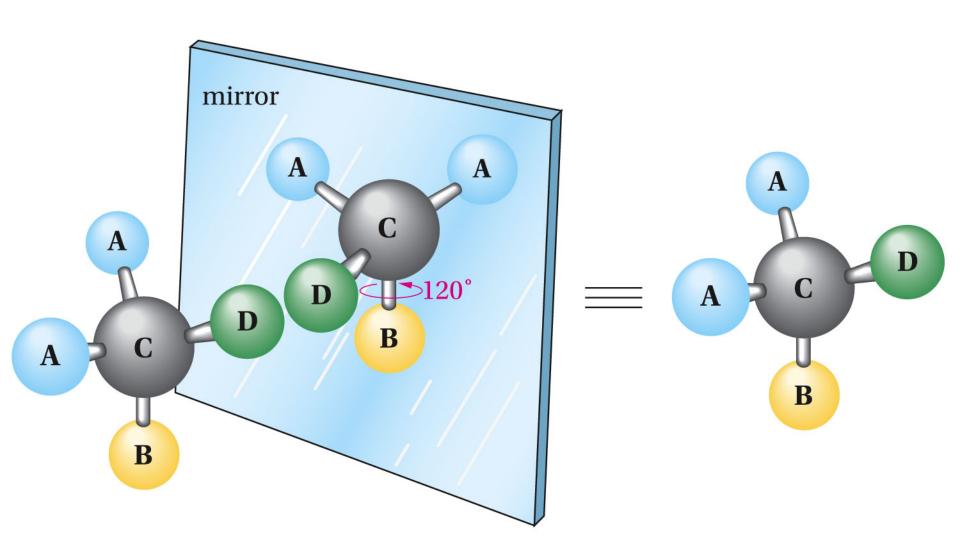
Stereogenic Centers: the Stereogenic Carbon Atom

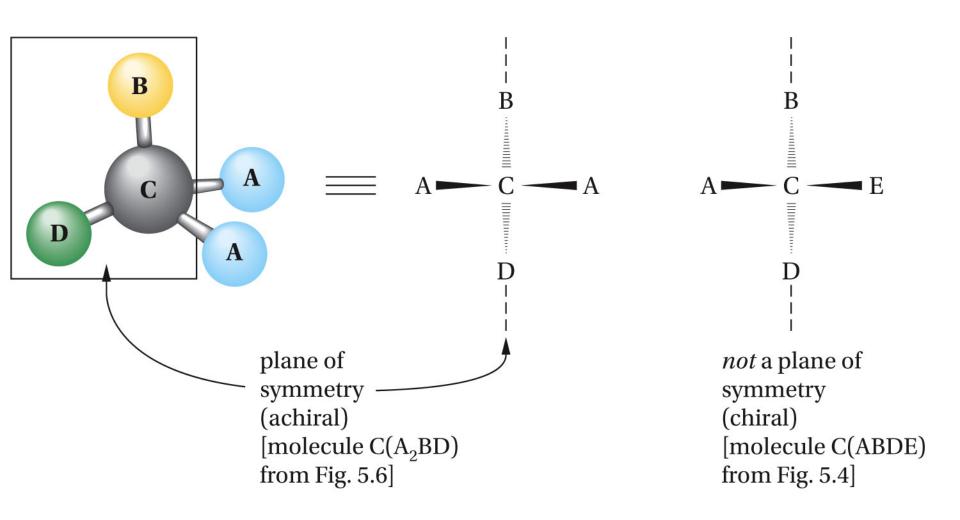
$$CH_3$$
 CH_3
 CH_3
 CH_2
 CH_3
 CH_3

Carbon atoms with four different groups attached to them are called stereogenic carbon atoms (also called chiral carbon)

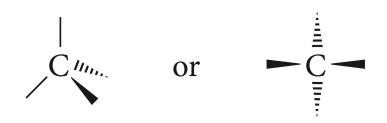


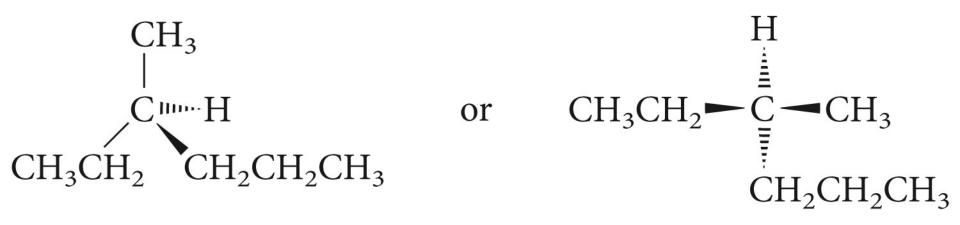




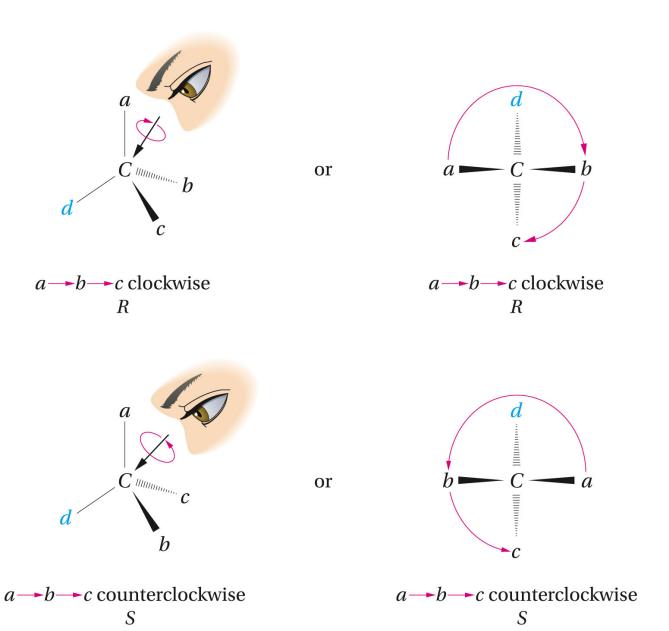


Locate the stereogenic center in 3-methylhexane and draw the two enantiomers of 3-methylhexane.





Configuration and the R-S Convention



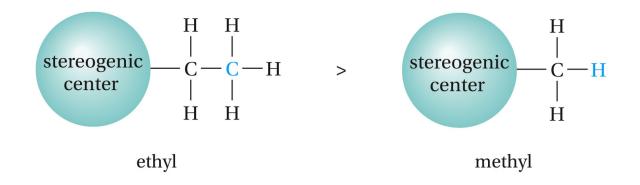
Rule 1

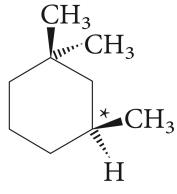
The atoms directly attached to the stereogenic center are ranked according to atomic number. The higher the atomic number, the higher the priority

$$Cl > O > C > H$$
high low
priority priority

Rule 2

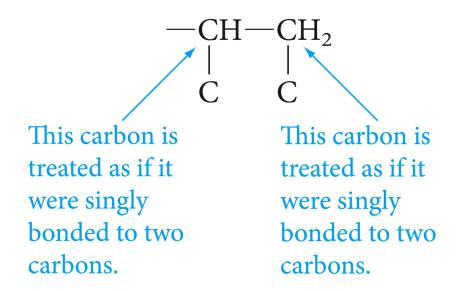
If a decision cannot be reached with rule 1, work outward from the stereogenic center until a decision is made. Example of ethyl and methyl below.





1,1,3-trimethylcyclohexane

Rule 3
Multiple bonds are treated as if they were an equal number of single bonds.



$$-C \equiv CH$$
 is treated as $-C - C - H$

$$-CH=O$$
 is treated as $-C-O$

Which group has the higher priority, isopropyl or vinyl?

$$\begin{array}{ccc} -\text{CH}(\text{CH}_3)_2 & \equiv & -\text{CH}-\text{CH}_2 \\ & \text{isopropyl} & & \text{CH}_3 & \text{H} \end{array}$$

Assign the configuration (R or S) to the following enantiomer of 3-methyl-hexane

