Stereochemistry and Chemical Reactions

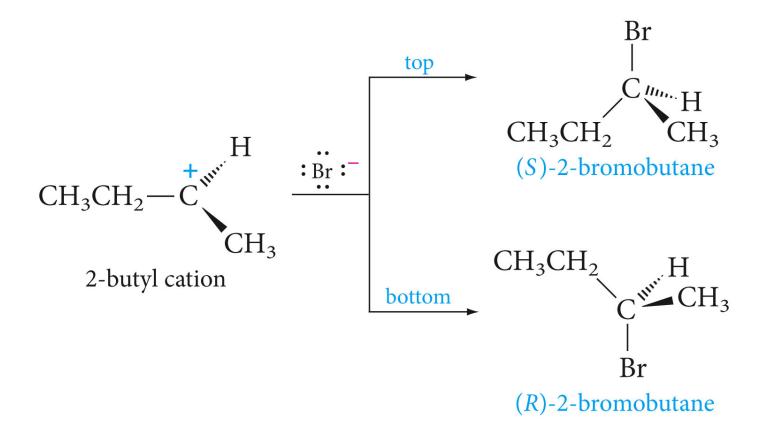
$$CH_3CH_2CH = CH_2 + HBr \longrightarrow CH_3CH_2\overset{*}{C}HCH_3$$

$$Br$$
1-butene 2-bromobutane

$$CH_{3}CH_{2}CH = CH_{2} + H^{+} \longrightarrow CH_{3}CH_{2}CHCH_{3} \xrightarrow{Br^{-}} CH_{3}CH_{2}CHCH_{3}$$

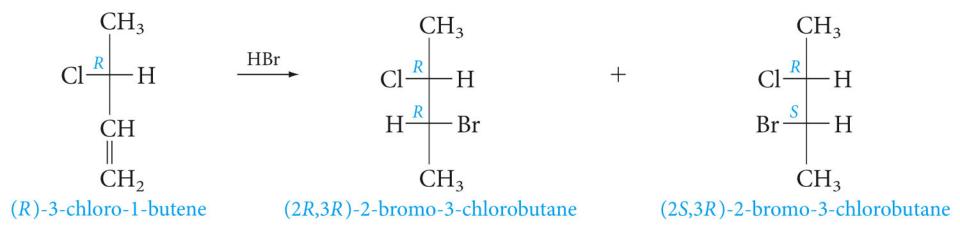
$$2-butyl \ cation$$

$$Br$$



When chiral products are obtained from achiral reactants, both enantiomers are formed at the same rates, in equal amounts.

Reaction of a chiral regent with an achiral reagent when it creates a new stereogenic center, leads to diastereomeric products at different rates and in unequal amounts.



Reaction of a chiral reagent with an achiral reagent, when it creates a new stereogenic center, leads to diastereomeric products at different rates and in unequal amounts.

Resolution of a Racemic Mixture

To separate a racemic mixture, we first react with a chiral reagent. The product will be a pair of diastereomers. These, differ in all types of physical properties and can therefore be separated by ordinary methods.

$$\begin{cases}
R \\
S
\end{cases} + R \longrightarrow
\begin{cases}
R - R \\
S - R
\end{cases}$$
pair of chiral diastereomeric enantiomers reagent products (separable)

(R)-asparagine

(S)-carvone

(R,R)-chloramphenicol

$$H$$
 OH $C(CH_3)_3$ N H $C(S,S)$ -paclobutrazol

(*R*)-thalidomide

$$H_3CO$$
 AcO
 RhL^*, H_2
 H_3CO
 AcO
 H_3CO
 H_3CO
 H_3CO
 H_3CO
 H_3O^+
 CO_2H
 H_3O^+
 CO_2H
 C